

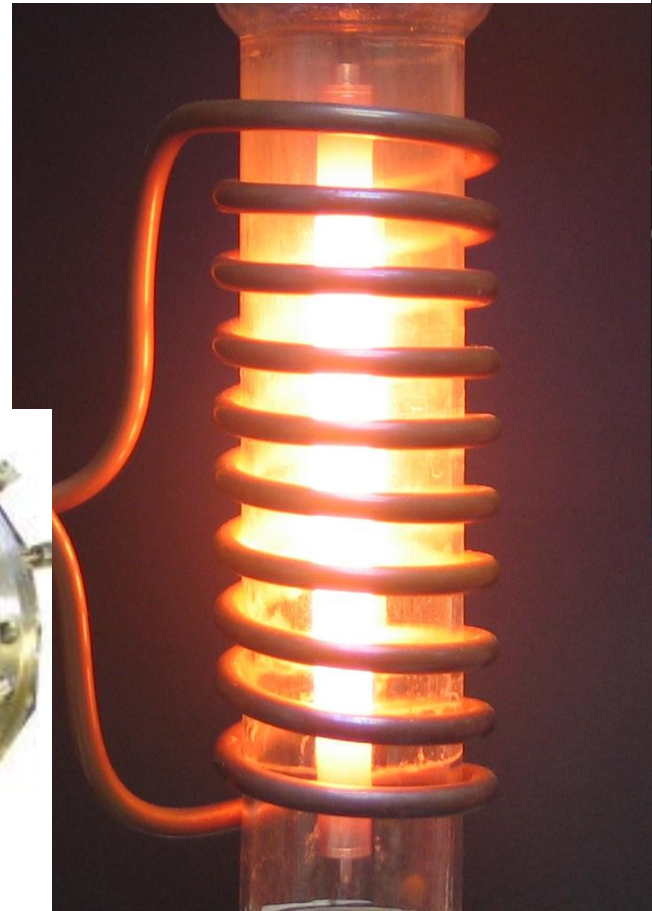
Status of studies on high-temperature oxidation and quench behaviour of Zircaloy-4 and E110 cladding alloys

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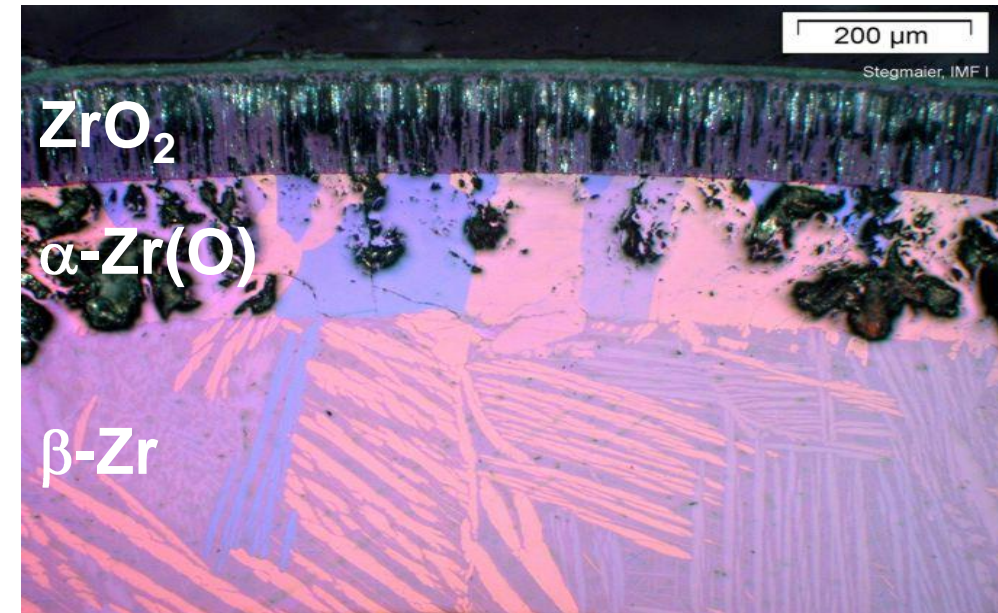
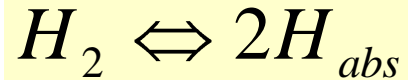
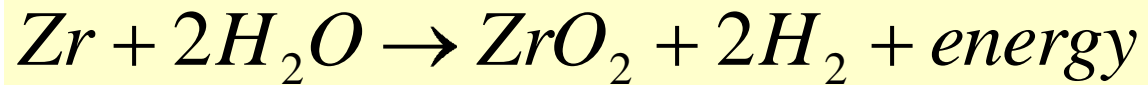
OUTLINE

- Introduction
- Separate-effects tests
- Bundle tests
- Modelling
- Future needs



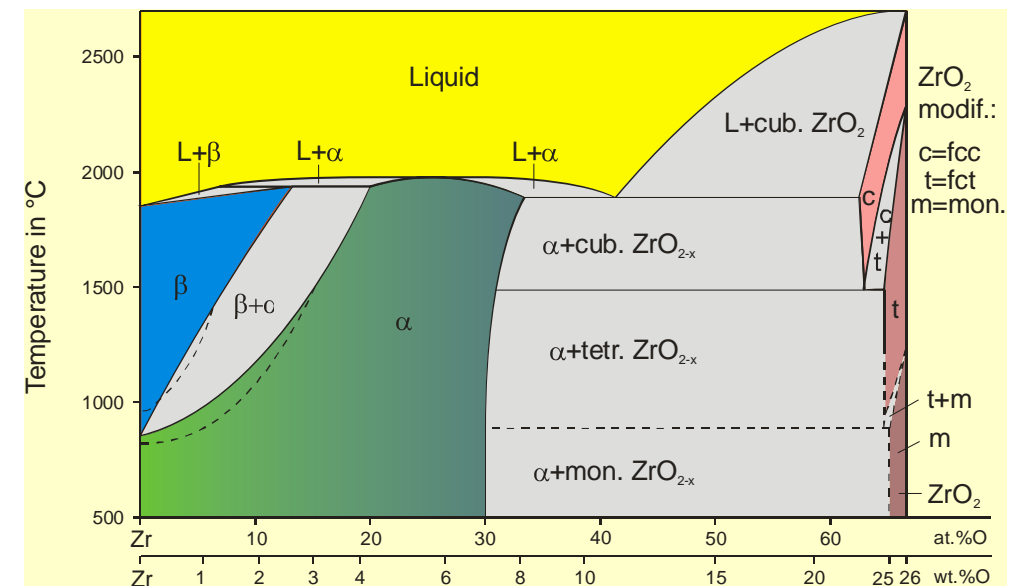
Introduction – Oxidation of Zr alloys

• Chemical reactions



• Diffusivity/solubility of gases

	Oxide	Metal	
		α	β
Oxygen	+++	++	+
Hydrogen	-	+	++



• Oxidation kinetics

$$\frac{\Delta m}{S} = k_m \cdot t^n$$

Mass gain

$$d_{\text{ZrO}_2} = k_o \cdot t^n$$

Oxide thickness

$$k = A \cdot \exp\left(-\frac{E_A}{RT}\right)$$

Temperature dependence

$n = 1/3$ (cubic) typical for low-temperature oxidation $< 900 \text{ }^\circ\text{C}$

$n = 1/2$ (parabolic) typical for high-temperature oxidation $> 1000 \text{ }^\circ\text{C}$

$n = 1$ (linear) typical for breakaway oxidation at $T < 1100 \text{ }^\circ\text{C}$
(after initial cubic or parabolic period)

Zircaloy-4:

- Used in western PWRs
- Zr production via chlorine chemistry and Kroll process
 - Sponge Zr
 - Impurities: Al, Mg, Ca

E110:





- Used in Russian VVER and RBMK reactors
- Zr production via fluorine chemistry and electrolysis
 - Impurity: F

Composition (wt-%):




	Nb	Sn	Fe	Cr	O
Zry-4	-	1.5	0.2	0.1	0.14
E110	1	-	0.008	0.002	0.05

Zr: balance

Separate-effects tests

- Isothermal tests (TG, furnace) in
 - oxygen 
 - steam 
 - air
- Transient tests (TG, furnace) in
 - oxygen
 - steam 
- Single rod quench tests with 
 - fresh samples
 - irradiated samples

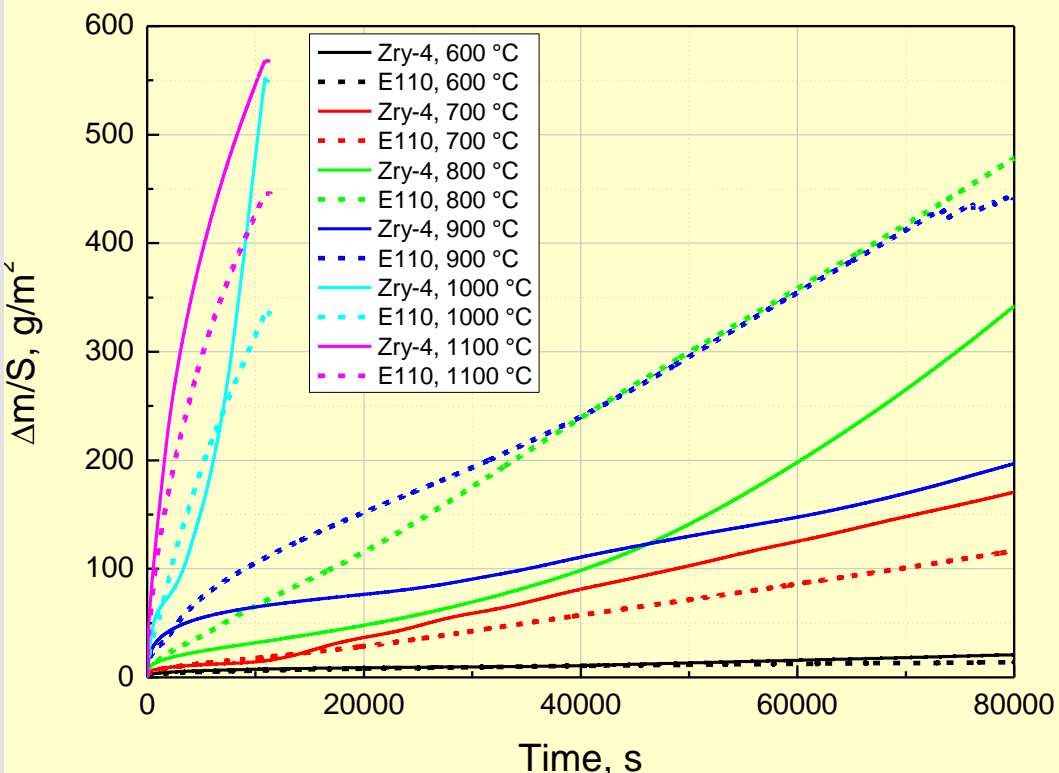
Bundle tests

- QUENCH-12 (vs. QUENCH-06) 
 - 31 rods
- PARAMETER-SF
 - 19 rods
- CODEX / CODEX-CT 
 - 7 rods
- PAKS cleaning tank incident 
 - 30 assemblies

this presentation 

Isothermal tests in oxygen

TG



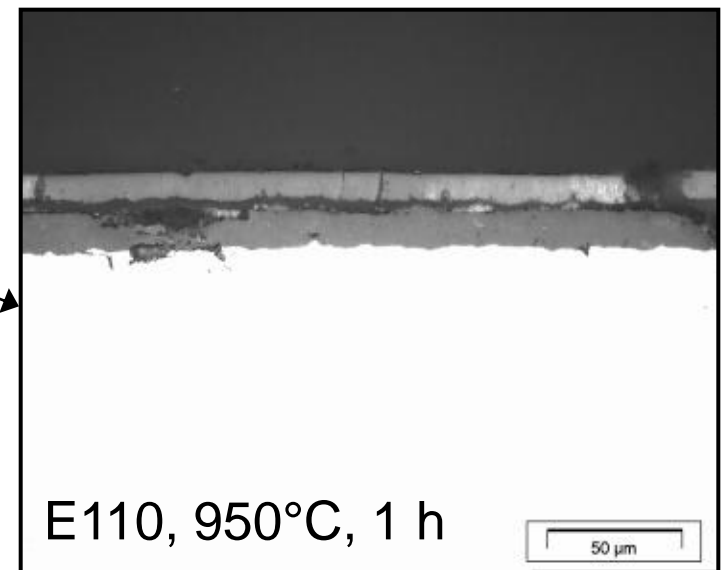
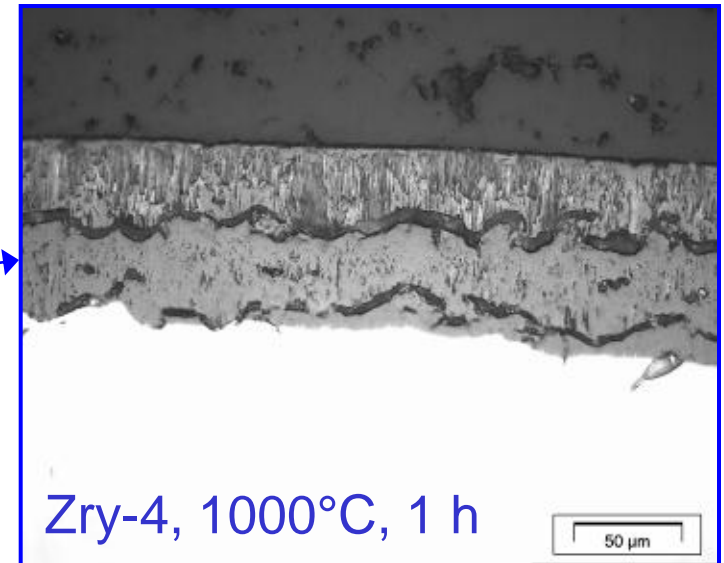
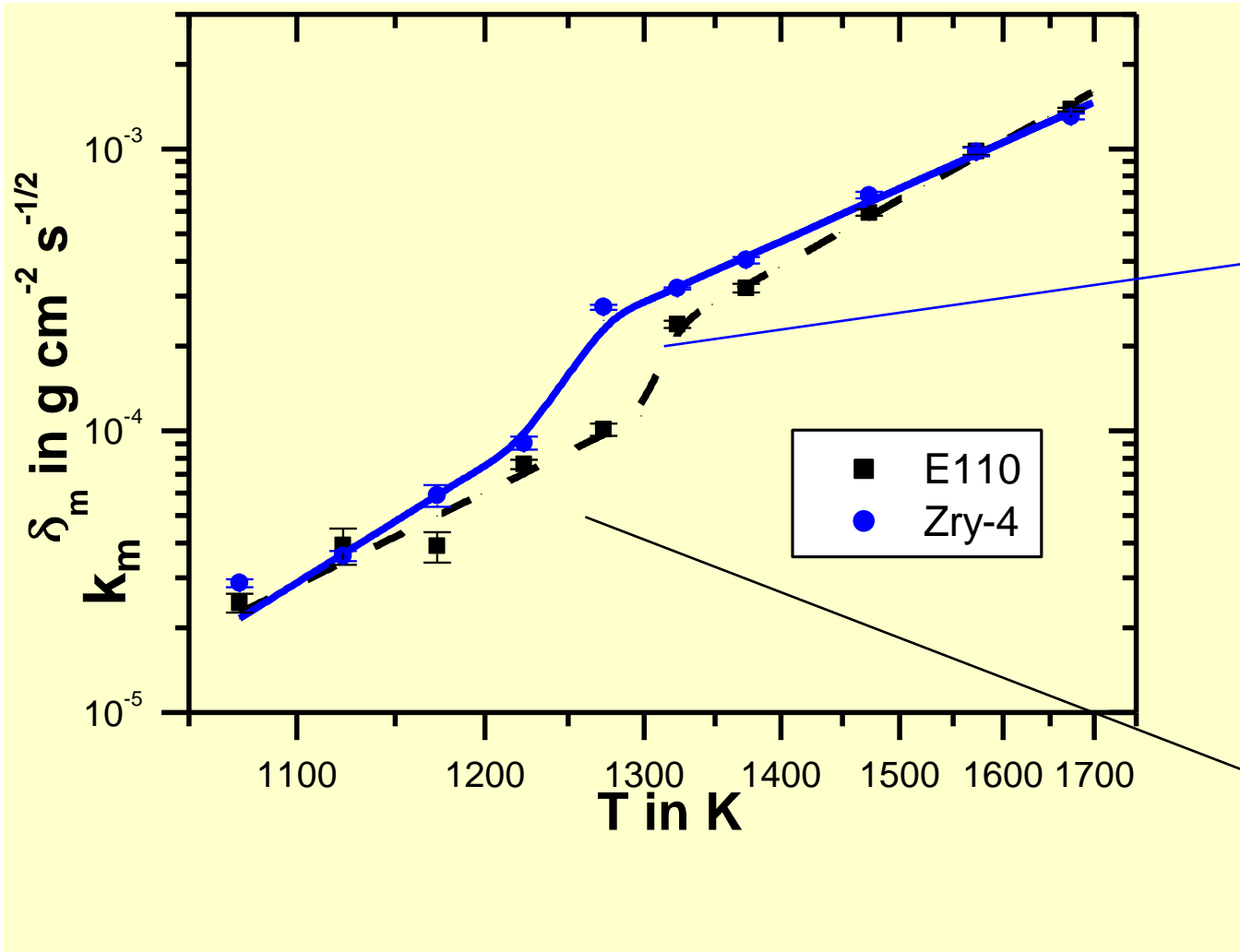
Microscopy

	Zry-4	E110
48 h 600 °C		
24 h 700 °C		
24 h 800 °C		
24 h 900 °C		
3 h 1000 °C		
3 h 1100 °C		

Post-test appearance

	48 h, 600 °C	24 h, 700 °C	24 h, 800 °C	24 h, 900 °C	3 h, 1000 °C	3 h, 1100 °C
Zry-4						
E110						

Isothermal tests in steam



Temperature dependence of mass gain rate

Transition to breakaway

ZrO ₂ , μm	Zry-4		E110	
T, °C	O ₂	Steam	O ₂	Steam
600	6	6	na	na
700	9	7	14	17
800	33	37	na	11
900	50	29	19	18
1000	69	75	30	43
1100	nt		nt	

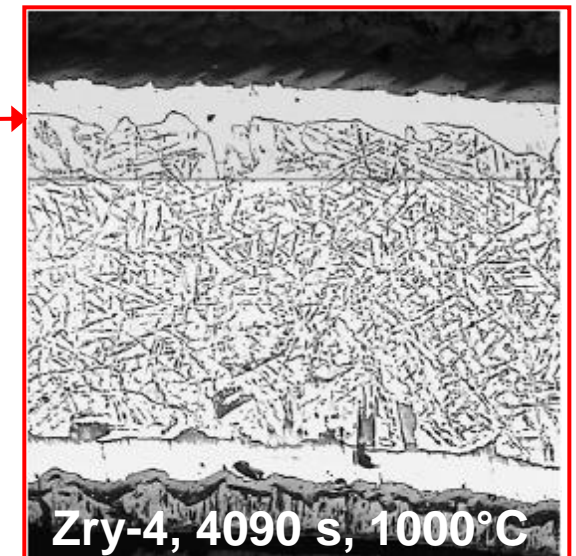
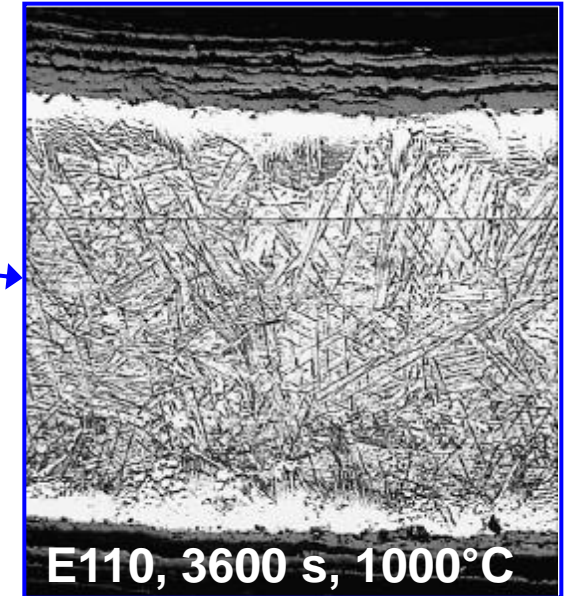
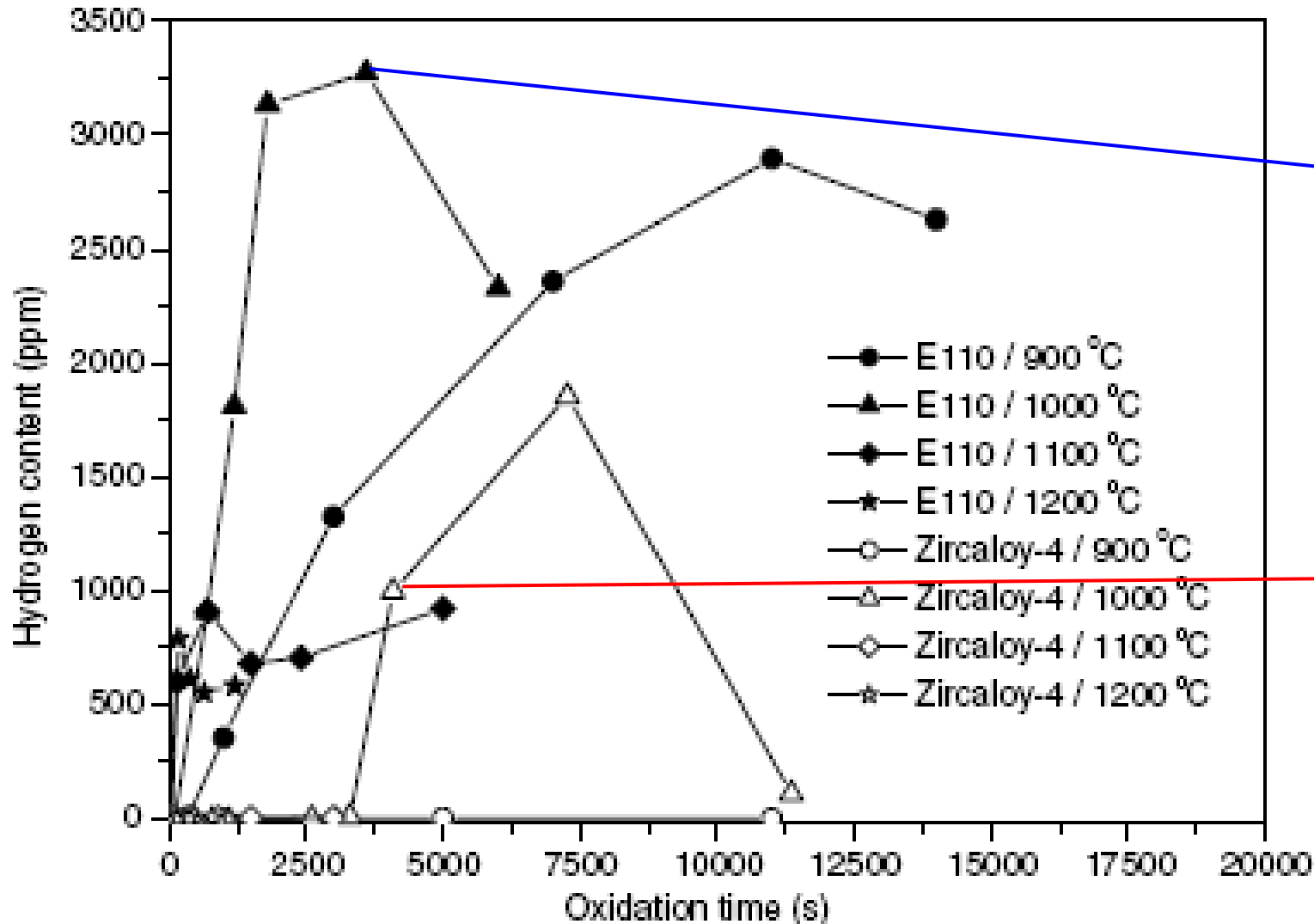
Oxide scale thickness (μm) at transition to breakaway

nt – no transition; na – not analysable



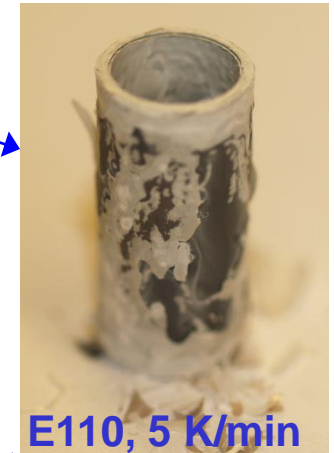
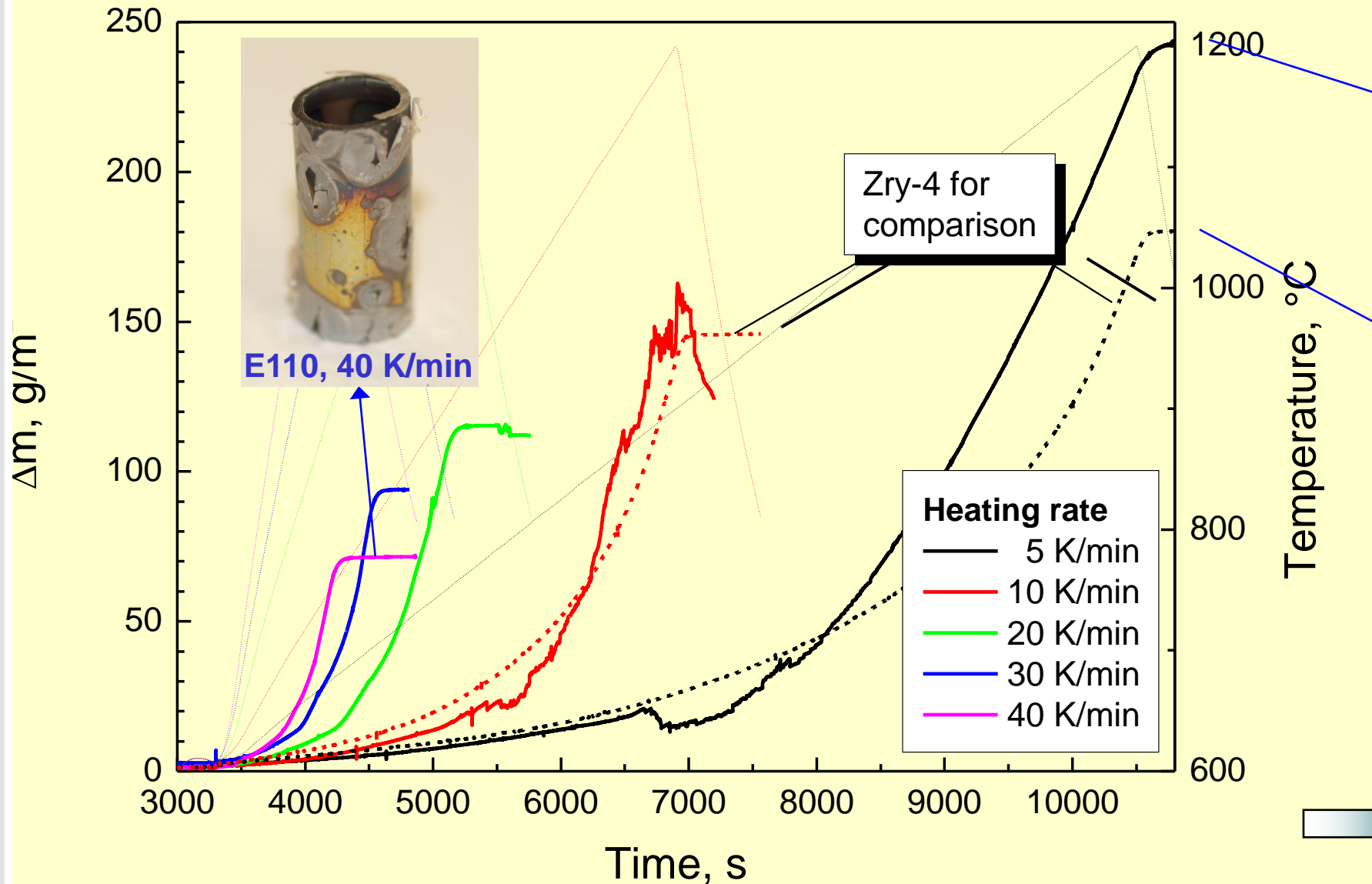
E110
24h, 800°C, steam

Hydrogen absorption during steam oxidation



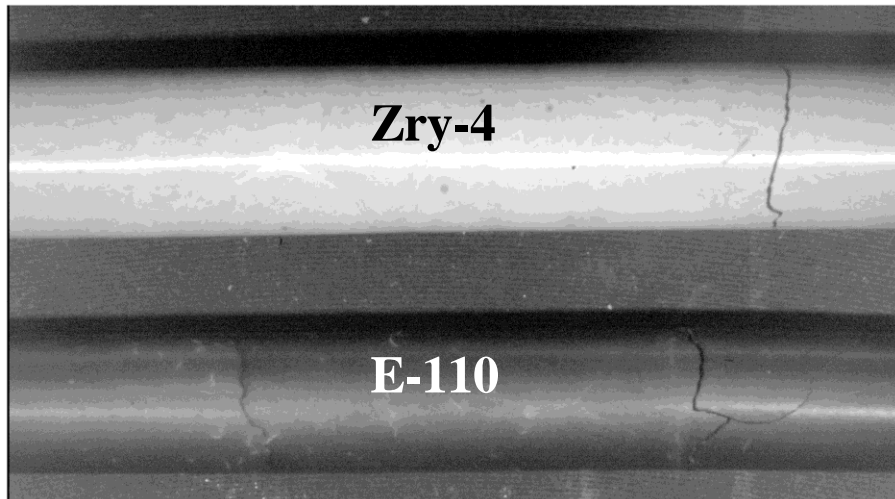
Hózer et al., JNM373(2008) 415

Oxidation under transient conditions



Transient oxidation of Zry-4 and E110 in steam in dependence on heating rate

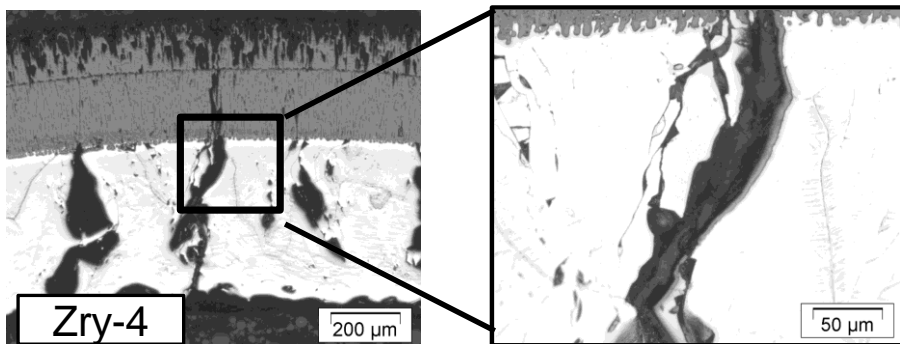
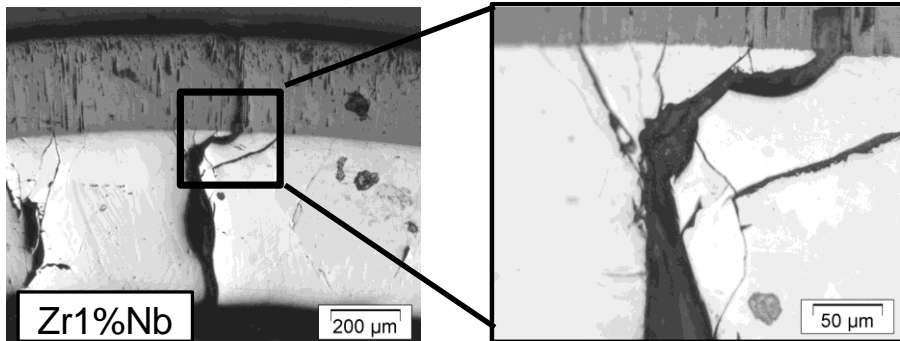
Single rod quench tests



- Different colours and therefore emissivity
- Oxidation of through-wall cracks at Zry-4, but not at E110
- Increased hydrogen absorption during quenching for Zry-4

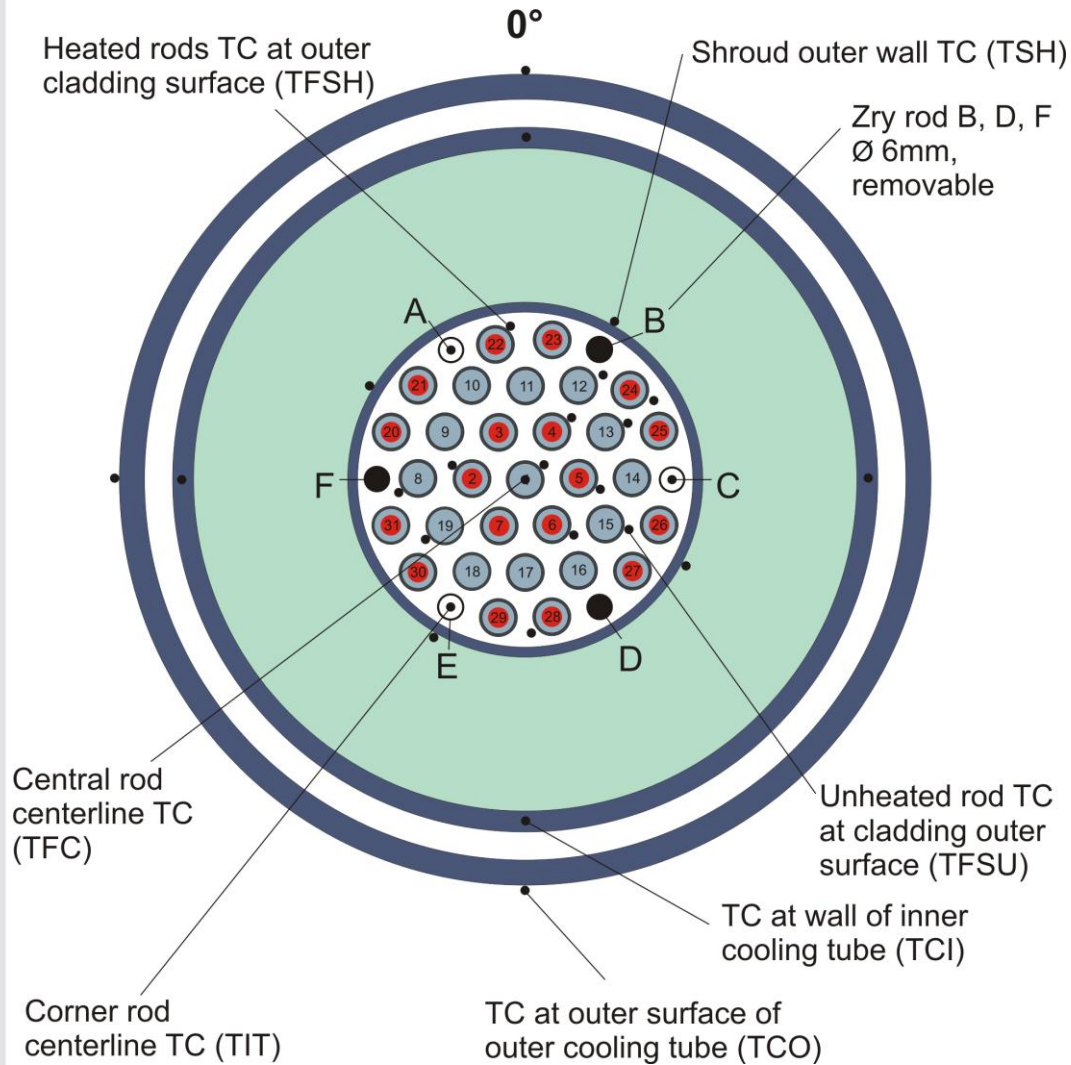
Irradiated simulators (E110):

- Earlier fragmentation and increased hydrogen release during oxidation
- Fuel swelling and clad creep causing contact between pellet and cladding lead to inner oxidation and embrittlement

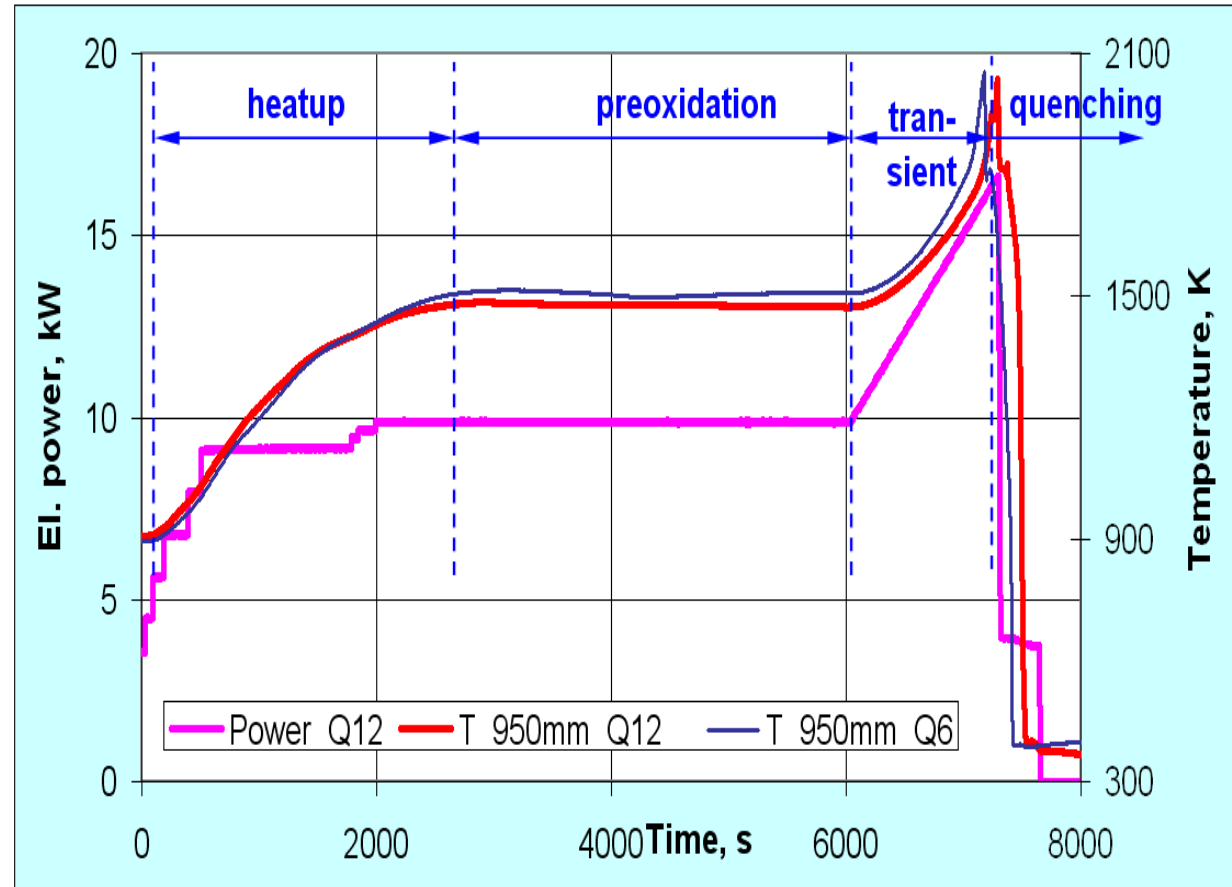


Oxidation and quenching from 1400°C

VVER bundle test QUENCH-12



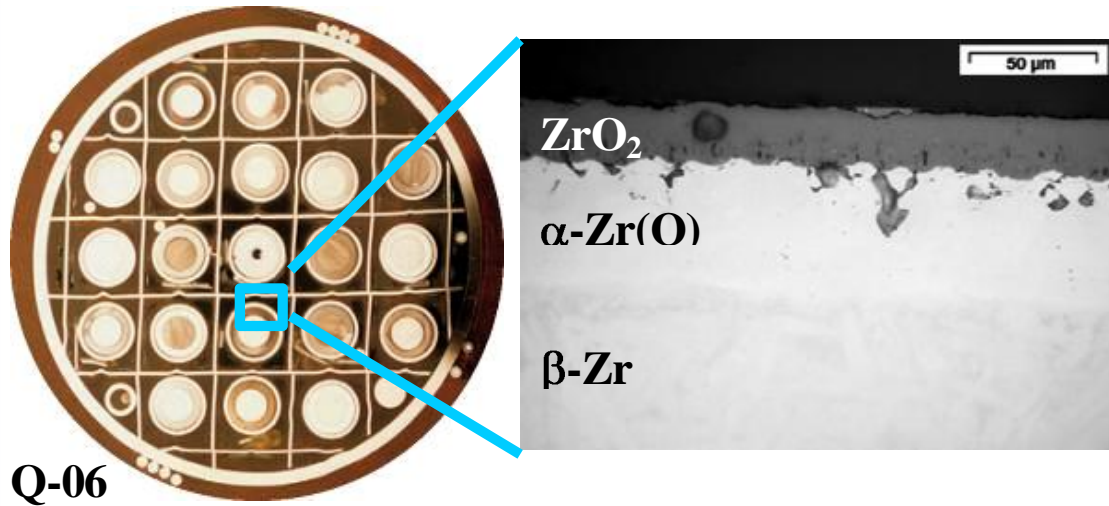
Bundle cross section



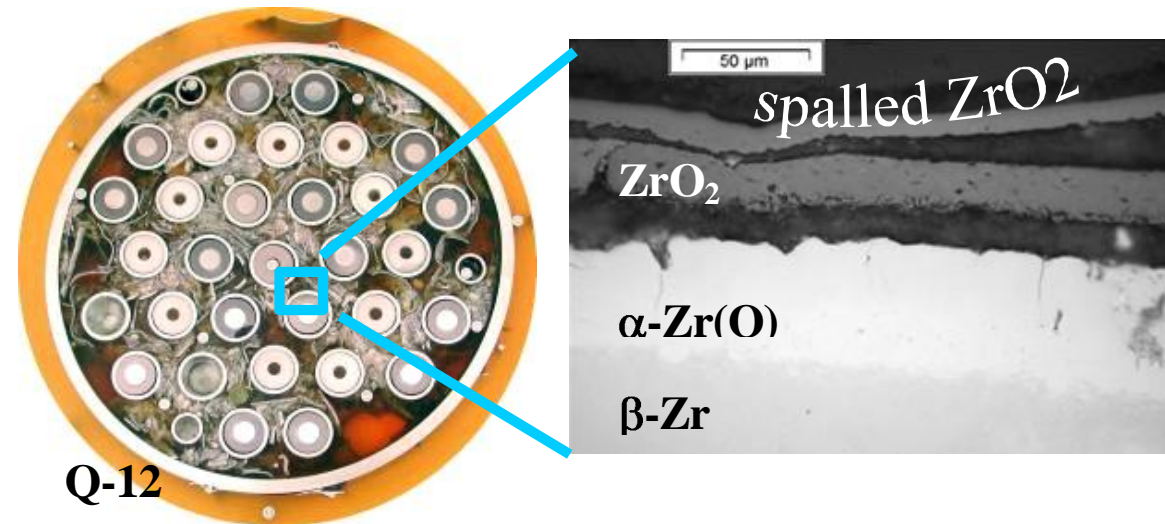
Power history and max. temperatures for QUENCH-12 and QUENCH-06

⇒ Very similar boundary conditions

QUENCH-12 post-test appearance



QUENCH-06: Zry-4



QUENCH-12: E110

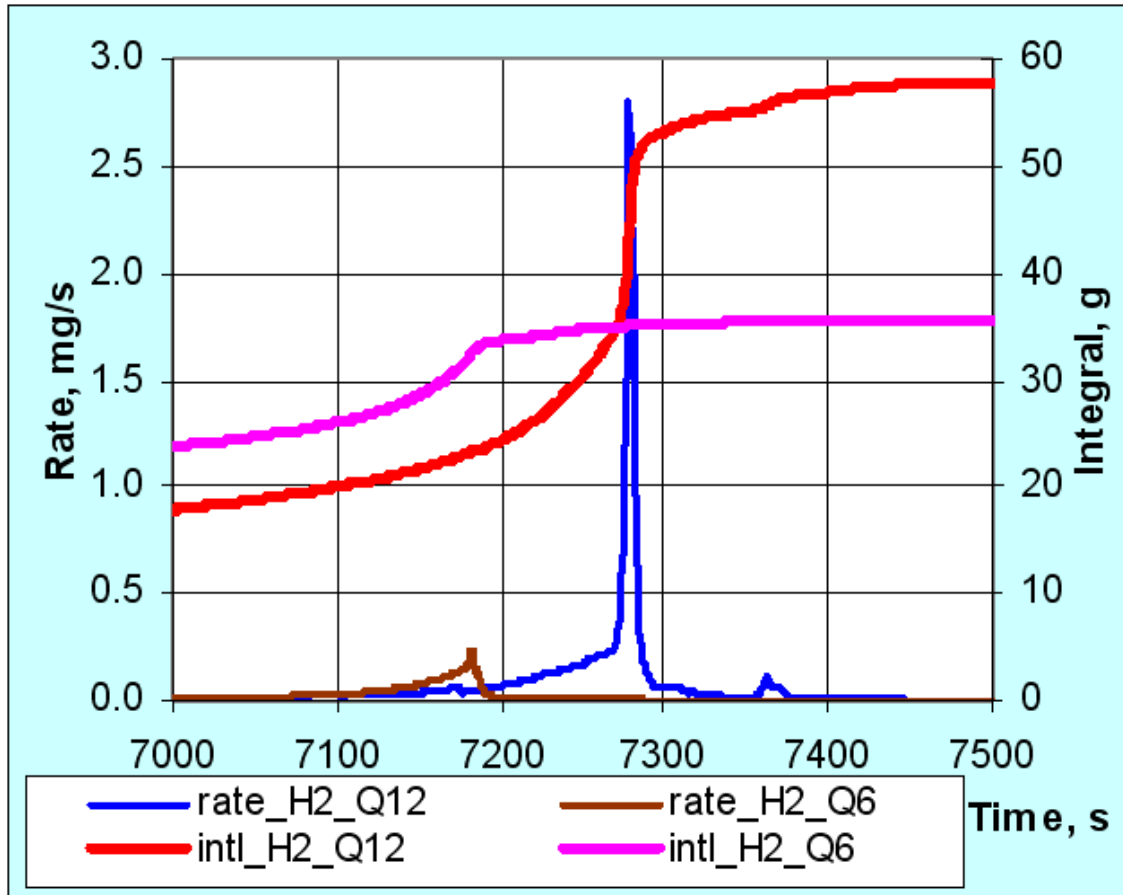
QUENCH-12:

- Intensive spalling of oxide scales
- Relocation of oxide debris to lower parts of the bundle

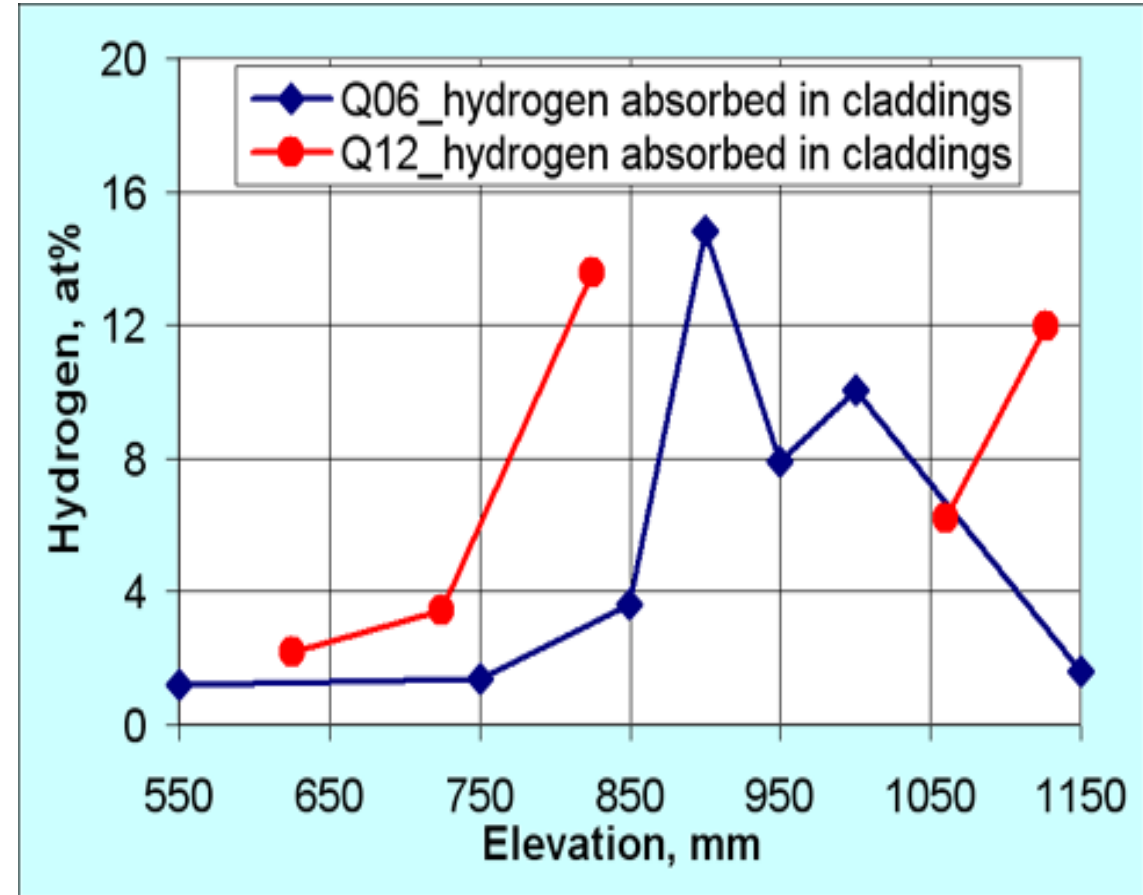


QUENCH-12 PT view

QUENCH-12 hydrogen (vs. QUENCH-06)



Hydrogen release



Hydrogen uptake

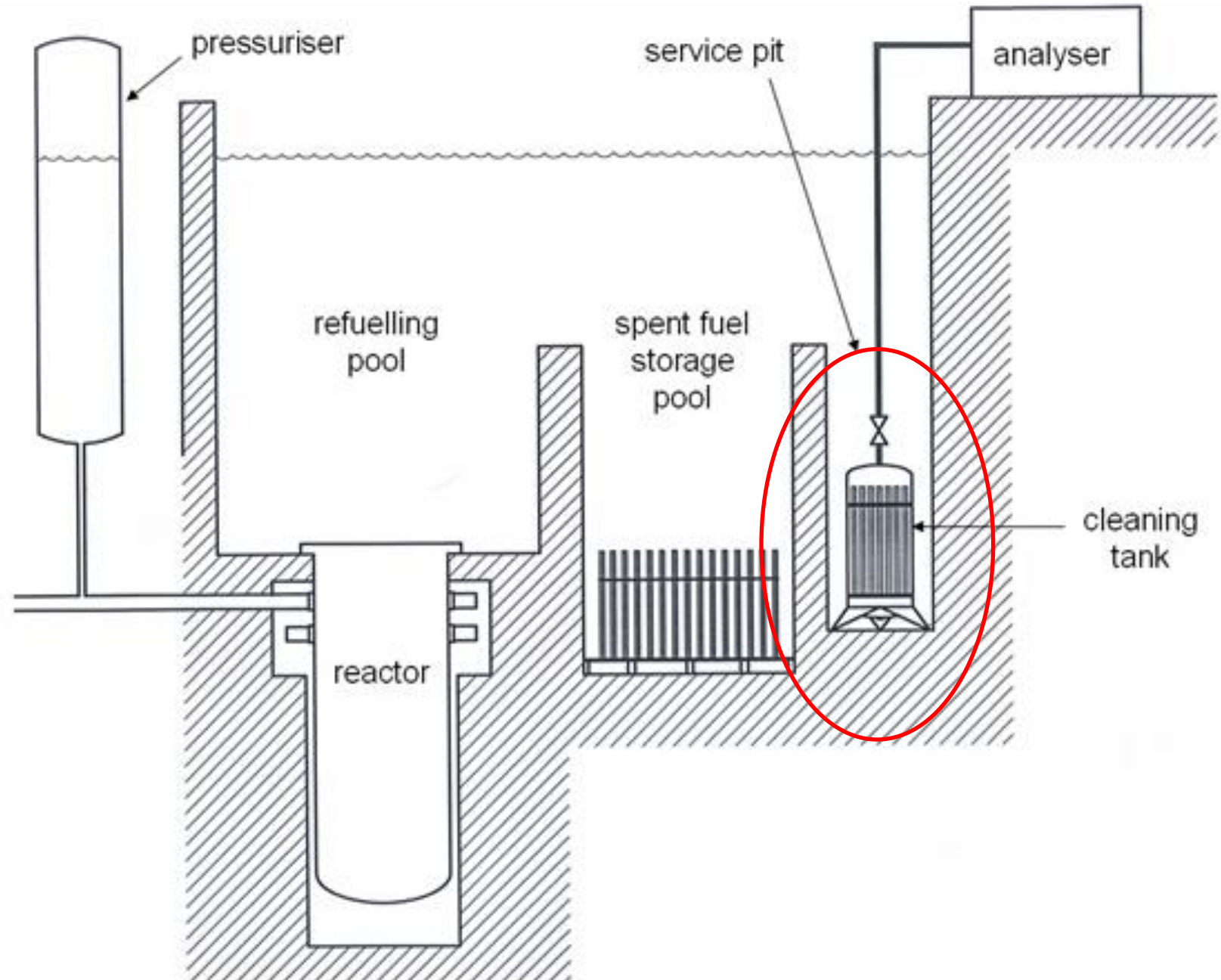
QUENCH-12:

- Higher hydrogen release during quench phase (24 g vs. 4 g in QUENCH-06)
- Increased hydrogen uptake by the remaining metallic cladding

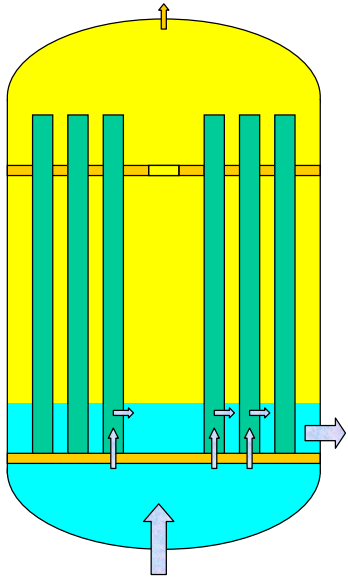
Paks-2 cleaning tank incident

High temperature oxidation of low power irradiated fuel assemblies for 7 hours in H rich steam

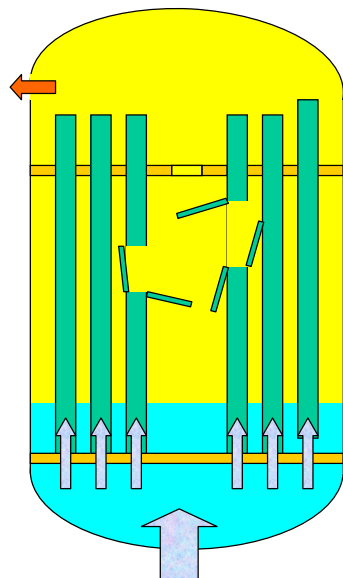
Fragmentation of fuel assemblies during opening of cleaning tank cover (quench by the water of pool)



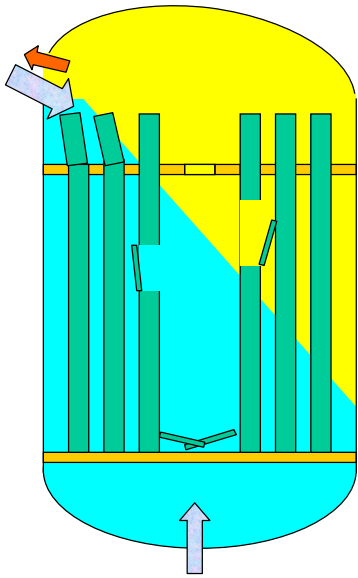
Paks-2 cleaning tank incident



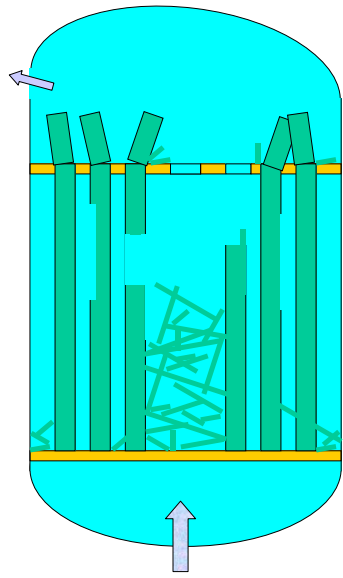
1) Insufficient cooling



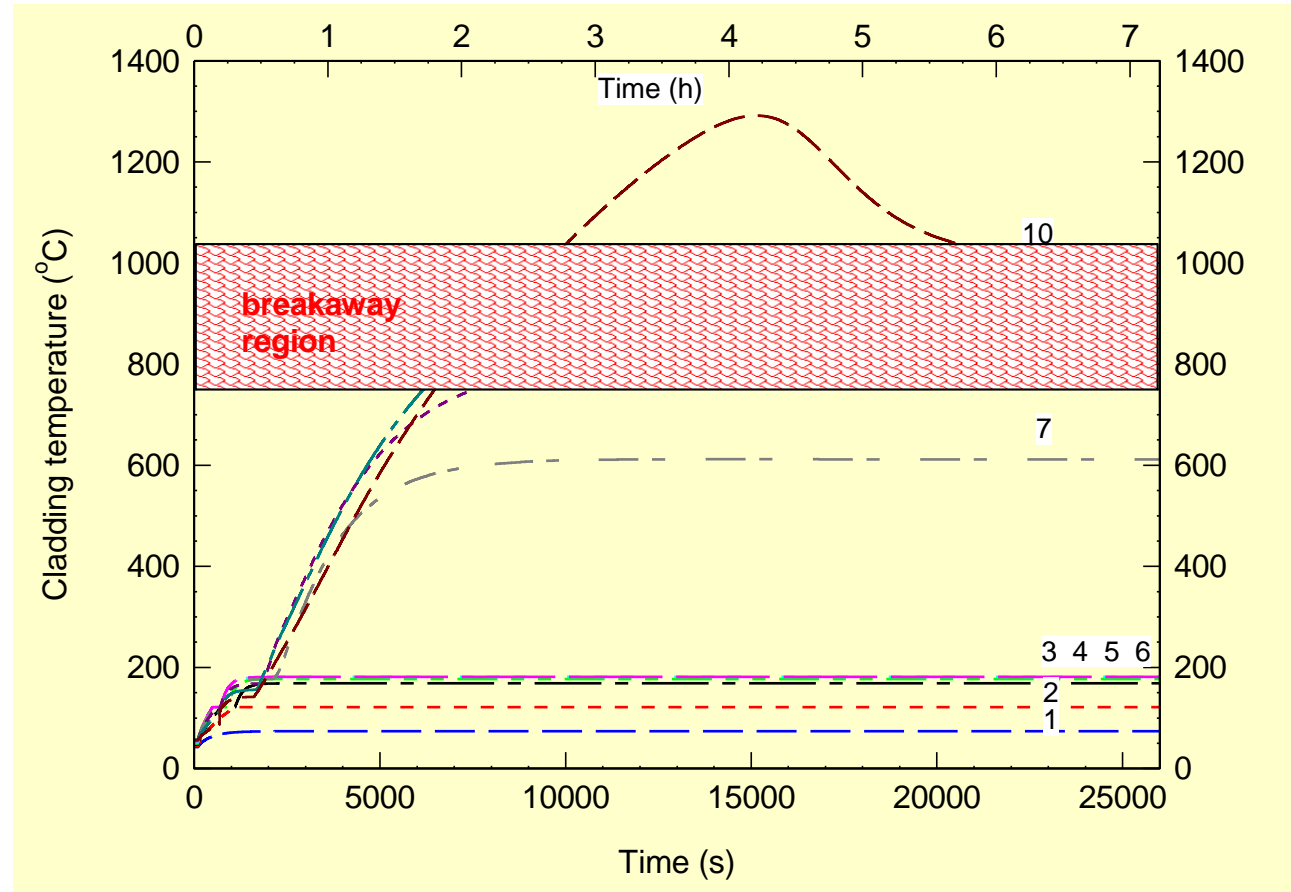
2) Ox. in steam/H₂



3) Flooding

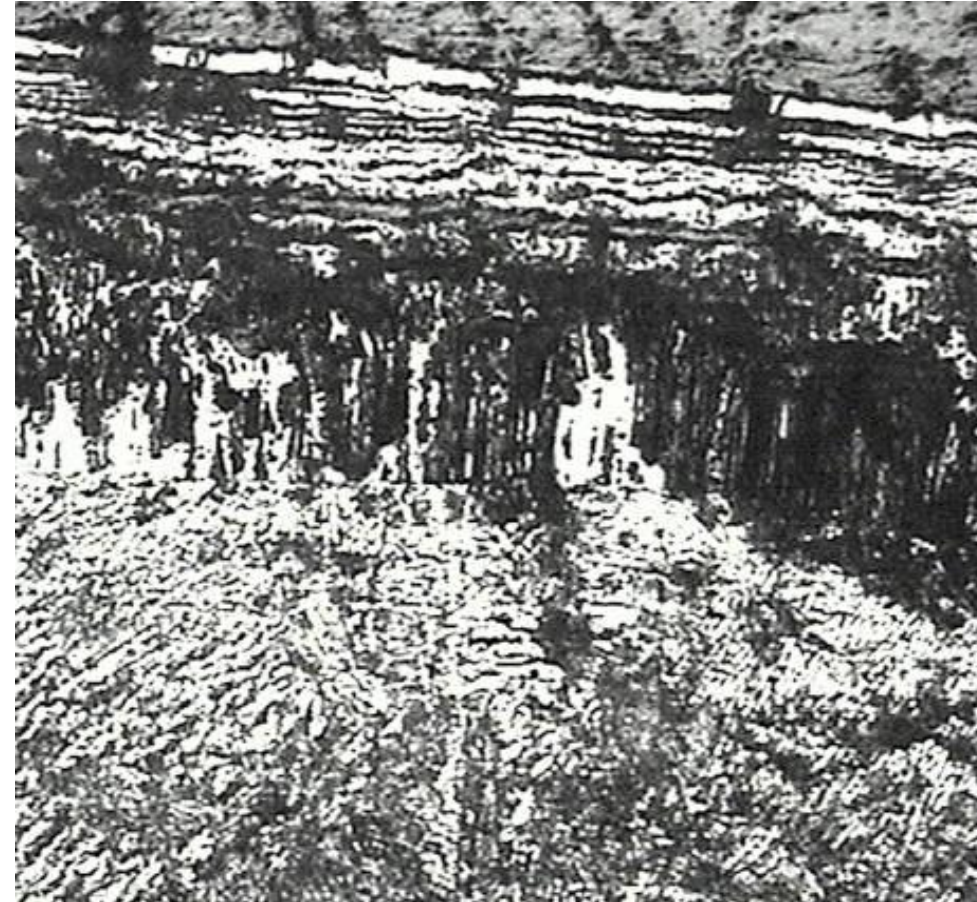
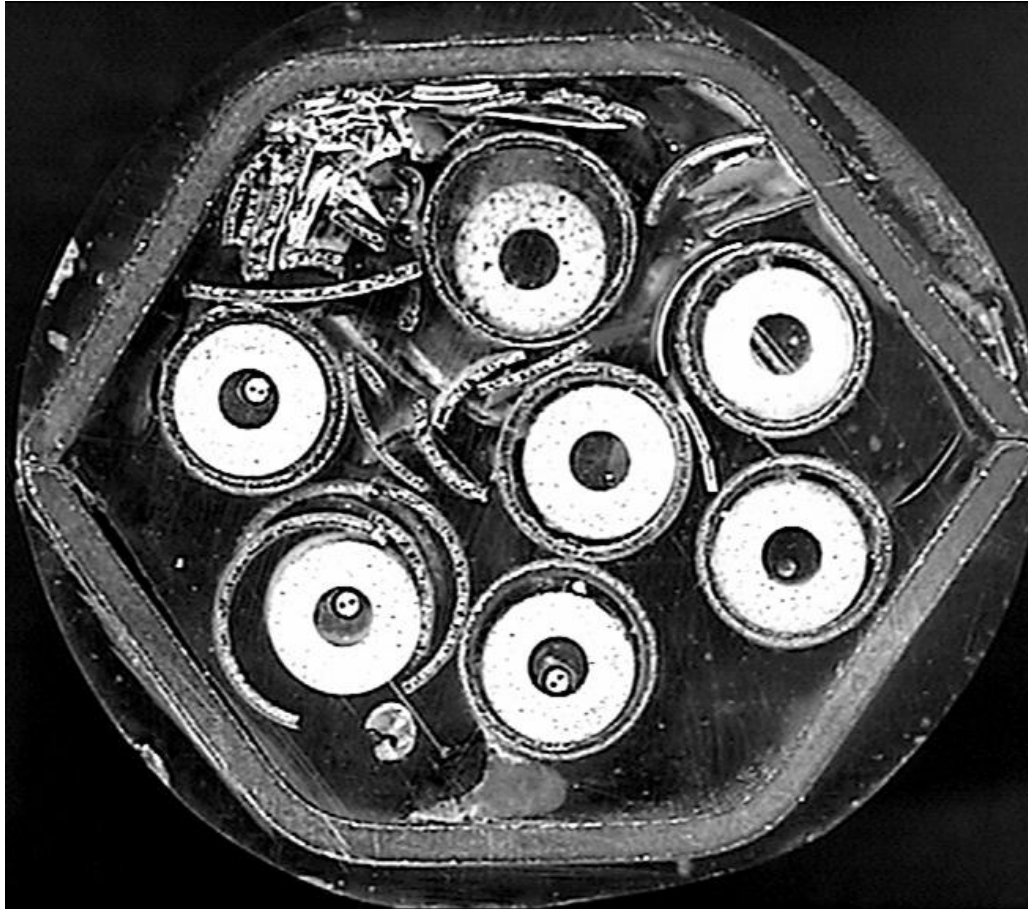


4) Collapse of FRs



Calculated temperature history

CODEX-CT PT appearance (Paks-2 simulation)



⇒ Ballooning and fragmentation of fuel rods

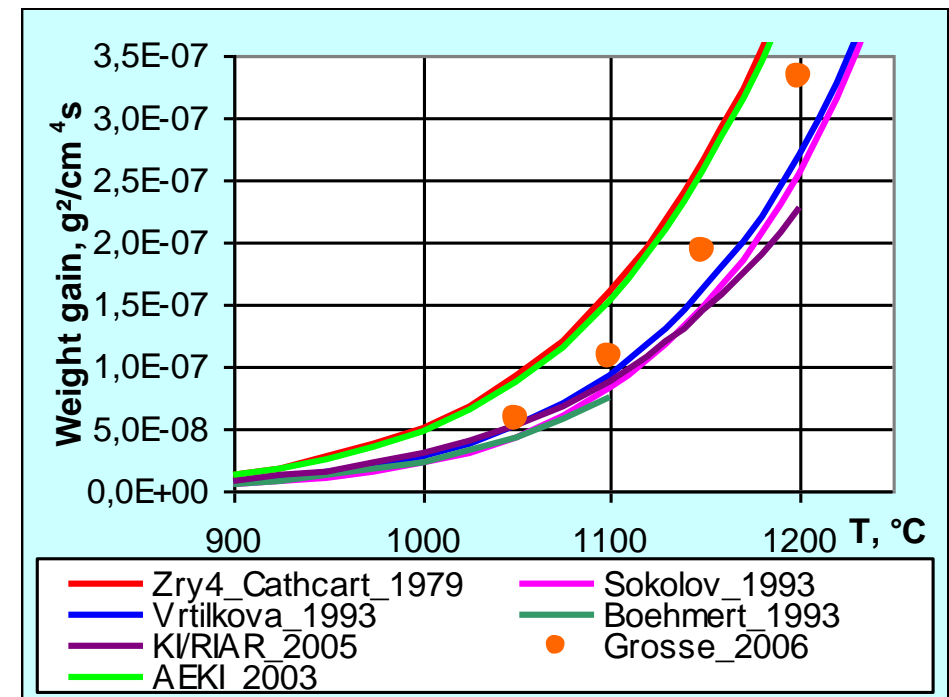
⇒ Formation of layered oxide scale on the cladding

⇒ High H absorption by Zr

Modelling of oxidation kinetics

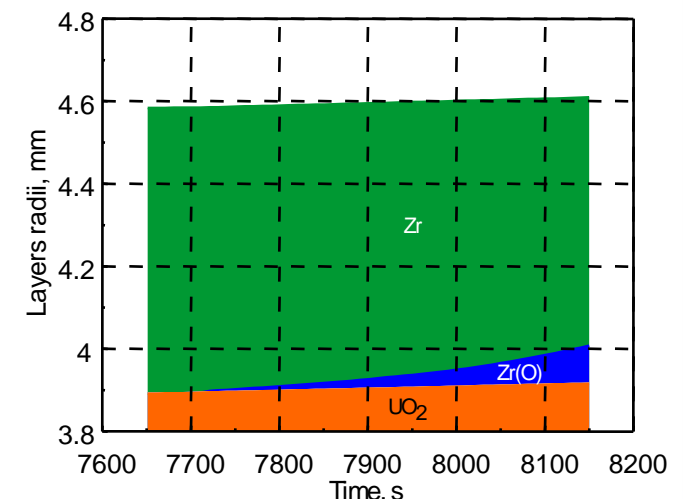
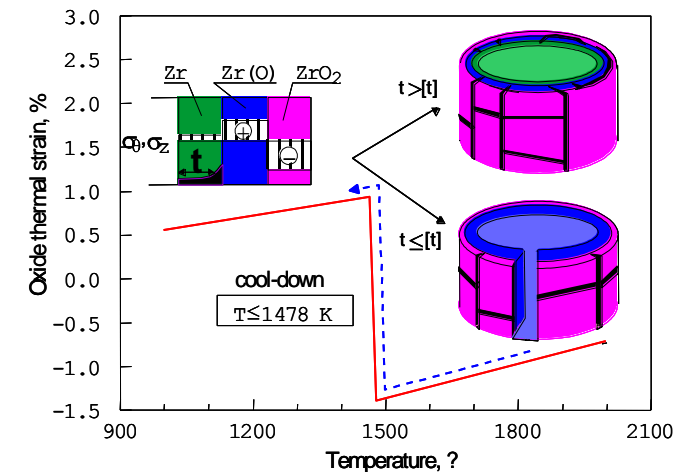
- Parabolic oxidation kinetics applied in all codes
- Critical assessment of all experimental data and best-fitted correlations available for Zircaloy-4 (Schanz, Veshchunov):
 - Cathcart-Pawel (or Leistikow-Schanz) in low-temperature range
 - Prater-Courtright in high-temperature range
- For E110 many data available, but comprehensive analysis still missing

Alloy	Kinetic	Temperature range, °C	Oxidation degree, μm	K_{ow} , ($\text{g}^2/\text{cm}^4\text{s}$)	T_{ew} , K
Zry-4	UH (AECL, 1978)	1580-1850	$\delta_{ox} > 200$	0.879	16610
Zry-4	PC (PNNL, 1986)	1630-2000	$100 < \delta_{ox} < 600$	32.95	26440
Zry-4	CP (ORNL, 1979)	1000-1500	$\delta_{ox} < 75$	0.3622	20100
Zry-4	LS (FZK, 1983) [38]	900-1500	$\delta_{ox} < 600$	0.52418	20962
Zry-4	FZK, 2006	1050-1400	$50 < \delta_{ox} < 550$	0.0888	18067
E110	VNIINM, 1988 [4]	550-1200	$\delta_{ox} < 400$	0.8464	20820
E110	VNIINM, 1993 [29]	1200-1500	$\delta_{ox} < 100$	1.59	23040
E110	VNIINM, 1993 [29]	1500-1600	$\delta_{ox} < 150$	0.9825	20800
E110	VV, 1993 [42]	600-1450	$\delta_{ox} < 250$	0.561	21426
E110	BDL (FZR) [3]	700-1100	$\delta_{ox} < 100$	0.2375	20522
E110	KI/RIAR, 2005 [5]	800-1200	$\delta_{ox} < 100$	0.09772	19105
E110	AEKI, 2003 [47]	500-1200	$\delta_{ox} < 150$	0.4330	20400
E110	FZK, 2006	1050-1400	$50 < \delta_{ox} < 350$	1.5952	22664



Further modelling activities

- Mainly in the frame of SVECHA/Q code development
- Thermo-mechanical behaviour of cladding during quenching
 - ⇒ Prediction of local (tensile and compressive) stresses causing deformation and rupture
- Influence of irradiation
 - ⇒ Successful application to RIAR SR quench tests with irradiated fuel rods (ISTC-1648.2)



- HT oxidation data of Zircaloy-4 and E110 have been compiled including SETs and integral bundle tests
- Oxidation is strongly dependent on temperature and atmosphere
- Integral and transient tests indicate a higher susceptibility of E110 to breakaway connected with less protective oxide scales, increased hydrogen absorption, and earlier embrittlement
- New data for model improvement are available

- Comprehensive analysis of high-temperature oxidation data of E110 is needed
- New development of E110 production procedure in Russia (change from “electrolytic” to “metal sponge” raw material) should be followed and the new material should be considered for HT investigations
- Advanced cladding alloys must be investigated carefully and oxidation model parameters have to be revised if necessary

Finally...

You are invited to the

14th International QUENCH Workshop

Forschungszentrum Karlsruhe (FZK)

November 4-6, 2008

www.fzk.de/quench

Thank you for your attention!